

RemoteALPHA - New developments in the metrology of environmental radiation monitoring of alpha emitters

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Motivation: Emergency Response Plans

Safety standards for the protection against the dangers arising from the ionising radiation: The European Directive 2013/59/EURATOM

Emergency Management System

Article 97

-Member states should ensure that account is taken of the fact that emergencies may occur in their territory...

-The emergency management system shall provide for the establishment of emergency response plans...

Emergency Preparedness

Article 98

-Member States shall ensure that emergency response plans are established in advance for the various types of emergencies...
-Member States shall ensure that emergency response plans are tested and revised at regular intervals...

International Cooperation

Article 99

Member States shall cooperate with other Member States and with third countries in addressing possible emergencies on its territory which may affect other Member States or third countries...

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Possible radiological emergency: Accidental or deliberate dispersion of alpha emitting radionuclides in the environment





DNA breaks caused by alpha, beta and gamma radiation.



Alpha Particles. Close Proximity Detection



http://www.argonelectronics.com/blog/the-value-of-applied-learning-forradiation-safety-training

Traditional detection methods (proportional counter, scintillator counter, PIPS detectors) are:

time consuming and tedious,

involve scanning very close to the surface of the contaminated area,

require the use **personal protective** equipment,

Expose the personel to other hazards and risks (other types of radiation, fire, etc.).

Motivation: Remote detection of alpha particles



Concept of remote detection of alpha particles.

Advantages:

- Operators are kept out of the radiation field,
- Efficient scanning of large areas.



Use of optical transitions in gas molecules: **radioluminescence**



Radioluminescence at a glance



Schematic representation of air ionization by α -particles.

High-energy alpha particles ionize air (predominantly molecular nitrogen).

Secondary electrons excite the air molecules, e.g.,

$$e^{-} + \mathrm{N}_{2}(\mathrm{X}^{1}\Sigma_{g}^{+}) \to \mathrm{N}_{2}^{*}(\mathrm{C}^{3}\Pi_{u}) + e^{-}$$
$$e^{-} + \mathrm{N}_{2}^{+}(\mathrm{X}^{2}\Sigma_{g}^{+}) \to \mathrm{N}_{2}^{*}(\mathrm{C}^{3}\Pi_{u})$$





Radioluminescence at a glance



Schematic representation of air ionization by α -particles and radioluminescence.

Air molecules emit fluorescent light (radioluminescence) in the UV range between 200 nm and 400 nm.

Range in air:

α-particles	\rightarrow	0,04 m
UV light	\rightarrow	500 m



Remote and real-time optical detection of alpha-emitting radionuclides in the environment



((((⊕ RemoteALPHA))))



Technical Workpackages

RemoteALPHA



source



Imaging of alpha emitters in the UVC (solar-blind) spectral range



Experiments at the University of Tampere (Finland), Research Group of Prof. Juha Toivonen

F. S. Krasniqi, T. Kerst, M. Leino, J.-T. Eisheh, H. Toivonen, A. Röttger, J. Toivonen, Nuclear Inst. and Methods in Physics Research, A **987** (2021) 164821







Schematic representation of the UV-C test setup.

UV-C radioluminescence: Detection with telescope and PMT





(a) A photo of the optical system for alpha particle detection. (b) Radioluminescence image of the Am-241 sample (32 MBq) in the UV-C spectral region.



Typical radioluminescence intensity distribution.



UV-C radioluminescence: Amplification with NO



By adding only 3 ppm NO to the air/N_2 amosphere, up to 500-fold increase of the radioluminescence signal.







Optimisation of optical configurations for detection of alpha-induced radioluminescence





Systems based on the reflective optics



Systems based on refractive optics (lenses)





Testing of the radioluminscence setup with **environmental samples**



well characterised **alpha-active environmental samples** concrete, mineral-phase, soil, organic and plant



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Thank you!





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UV-C radioluminescence: interference from sunlight is supressed



D. W. Wilmouth et. al., Green Chemsitry, Ch. 3.3 (2018), https://doi.org/10.1016/B978-0-12-809270-5.00008-X



a) T. Kerst und J. Toivonen, Optics Express **26**, 33764 (2018), und b) S. M. Baschenko, J. Radiol. Prot. **24**, 27 (2004)



Optimization of the optical system



Main optimization parameters:

Enlargement of the receiving optics

 (e.g. from Ø 100 mm to more than 300 mm)
 Change of the arrangement of the optical elements (lenses and mirrors)



Resesarch on radioluminescence amplification Dependence on gas atmosphere Chamber with quartz-window Alphasource



UV-C radioluminescence: Amplification with N₂



Flooding the chamber with N_2 , at a rate of 5 L/min increases radioluminescence by about a hundredfold compared to air.



Experimental setup with CCD



UVC radioluminescence image of a 3.7 MBq source (right) and a photo of the sample in the experimental chamber (left).



RemoteALPHA: Workpackages

WP1

To develop a new method and instrumentation for the optical detection of alpha particle emitters in the environment by air radioluminescence over a detection range of more than two metres. This includes the development of the first prototype of a mobile-outdoor optical detection system for real-time radioluminescence mapping of alpha sources in the environment.



To develop and establish a calibration system for the novel-type radioluminescence detector systems. This includes a new metrological infrastructure with a dedicated UV radiance standard, well characterised alpha-active environmental sample (mineral-phase, soil, organic and plant specimen spiked with alpha emitters) and a validated calibration scheme for the remote detection of optical system.



To extend the optical detection system to an imaging functionality for mapping of alpha contaminations in the environment. This includes the development of an unmanned airborne monitoring system (UAMS) that will integrate the unmanned aerial vehicle (UAV) and the novel alpha-radioluminescence detection system developed in the objective 1 to scan and obtain an image of the contaminated area.

WP4

To prepare and run a feasibility study for a laser-induced fluorescence spectroscopic method for the detection of alpha emitters. This method complements alpha-radioluminescence and, depending on laser parameters such as pulse power, photon wavelength and pulse duration, can enhance the detectable activity limit to below 1 kBq/cm2.

PB Optical Systems for Remote Alpha-Monitoring



Schematic of a Cassegrain telescope. By folding the light path one can reduce considerably the physical length of the telescope tube.

Schematic of a Newtonian-type telescope. Flexible optical system.

PB Laser Lab Tests: Fluorescence/Raman LIDAR

Supercontnuum fiber laser Power: ~10 W Rep rate: 20-320 MHz Spectral coverage: 400 - 2400 nm



Fluorescence/Resonant Raman Method at a Glance



1) Alpha particles ionize N₂ molecules.

2) Pulsed laser excites N₂⁺ ions at λ_{las} .

3) Fluorescence is emitted at $\lambda_{emission} \neq \lambda_{las}$

> Laser wavelength: 391 nm Energy per pulse: 500 µJ Rep. Rate: 10 Hz



PB Laser induced fluorescence in ambient air (corona discharge)

Ref.: Konthasinghe et al., Applied Spectroscopy 69, 1042 (2015)



Energy Levels of N₂ Molecule







Schematic of nitrogen ionization by α -particles.



Major constituents of dry air, by volume

Name	Formula	Volume in %	
Nitrogen	N ₂	78.084	
Oxygen	O ₂	20.946	
Argon	Ar	0.9340	
Carbon dioxide	CO ₂	0.041332	
Neon	Ne	0.001818	
Helium	Не	0.000524	
Methane	CH ₄	0.000187	
Krypton	Kr	0.000114	
Not included in above dry atmosphere:			
Water vapor	H ₂ O	0-3%	



https://en.wikipedia.org/wiki/Atmosphere_of_Earth



Number densities of air constitents:

 $\begin{array}{ll} n_{air} &= 2.65 \times 10^{19} & {\rm molecules/cm^3} \\ n_{\rm N2} &= 2.07 \times 10^{19} & {\rm molecules/cm^3} \\ n_{\rm O2} &= 5.55 \times 10^{18} & {\rm molecules/cm^3} \\ n_{\rm Ar} &= 2.48 \times 10^{17} & {\rm atoms/cm^3} \\ n_{\rm CO2} &= 8.75 \times 10^{15} & {\rm molecules/cm^3} \end{array}$

Number density of 1ppm NO: $n_{NO}=2.47 \times 10^{13}$ molecules/cm³

Molar mass of NO: *M*_{NO}=14.007 g/mol+15.999 g/mol=30.006 g/mol

Separation between air molecules

$$d_{air} = \left(\frac{3}{4\pi n_{air}}\right)^{1/3}$$

d_{air}∼2 nm

Radius of a molecule $r \sim 2x$ Van der Waals radius For N₂, $r \sim 2 \times 0.155$ nm=0.31 nm

Air molecules occupy only $(r/d_{air})^{3}=0.4\%$ of the volue they are spread.

1 ppm NO =1230 µg/m³ =1.23×10⁻⁹ g/cm³ After http://www.euro.who.int/ data/assets/pdf file/0017/123083/AQG2ndEd 7 1nitrogendioxide.pdf **PB** NO_x concentration



The German city of Stuttgart, for instance, has recorded levels of 82 micrograms of NO2 per cubic meter of air — that's twice as much as the allowed threshold of 40 micrograms. [After https://www.dw.com/en/eu-takes-germany-to-court-over-air-pollution/a-42351552]







Relative excitation cross section for the $C^3\Pi_u \rightarrow B^3\Pi_g$ band in N_2 . After J. T. Fons *et al.*, Physical Review A **53**, 2239 (1996).

PB Remote Detection of Alpha Particles: Resonant Raman LIDAR

